

FIG.1

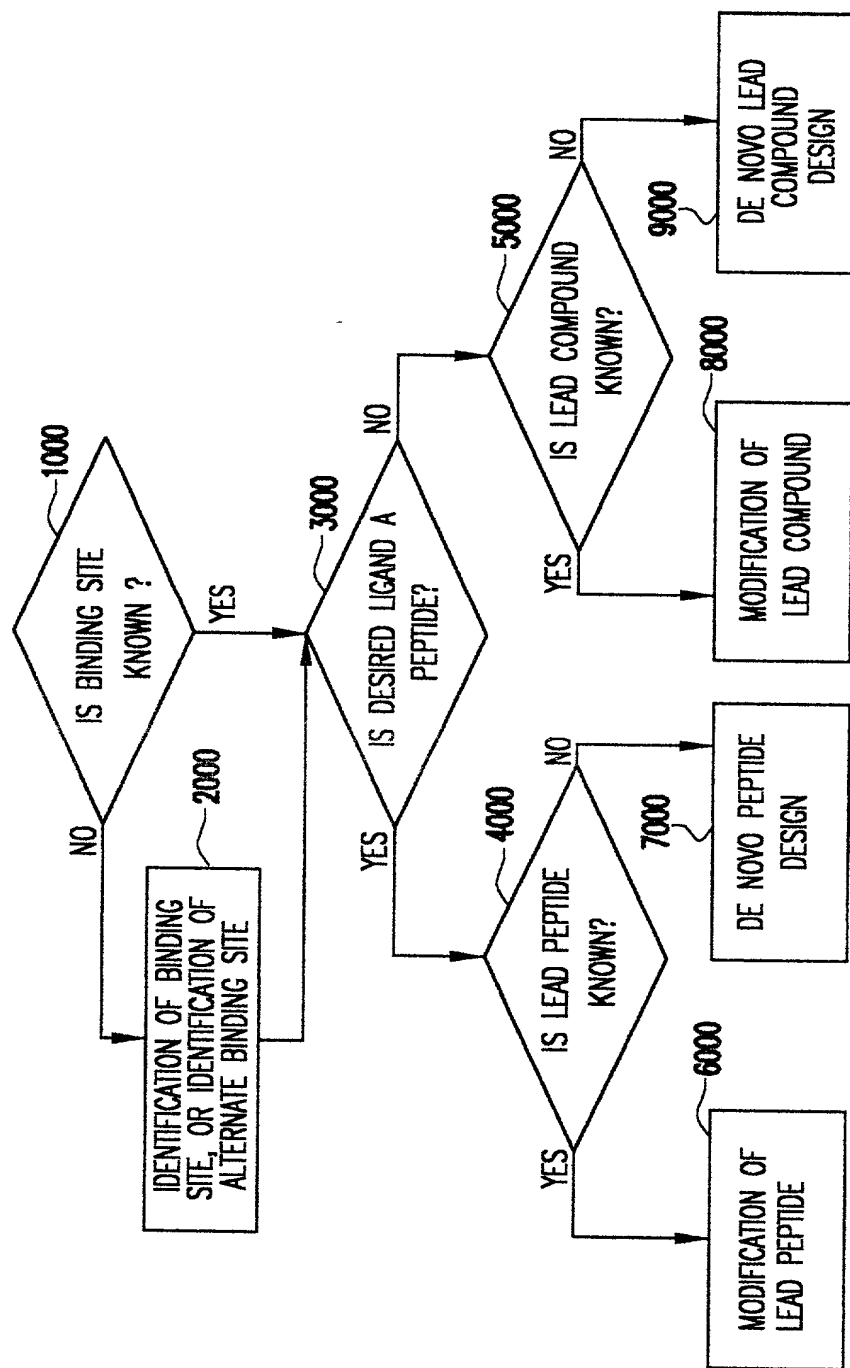


FIG. 2

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graph TD; 2100[INPUT ATOMIC COORDINATES OF ALL ATOMS IN TARGET MOLECULE] --> 2200[CALCULATE BINDING POTENTIAL OF EACH ATOM IN TARGET MOLECULE]; 2200 --> 2300[IDENTIFY REGIONS WITH HIGHEST BINDING POTENTIAL]; 2300 --> 2400[IDENTIFY POTENTIAL BINDING SITES IN TERMS OF BINDING POTENTIAL]; 2400 --> 2500[CLASSIFY POTENTIAL BINDING SITES ACCORDING TO CHEMICAL NATURE (E.G., DEGREE OF HYDROPHOBICITY, CHARGE, ETC.), SURFACE AREA, LOCATION IN CAVITY, ETC.]; 2500 --> 2600[SELECT POTENTIAL BINDING SITE FOR LIGAND DESIGN ACCORDING TO USER CRITERIA, E.G.:  
-TYPE OF LIGAND (E.G., PEPTIDE, ORGANIC MOLECULE)  
-BINDING AFFINITY  
-SIZE  
-GEOMETRY];
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2100 INPUT ATOMIC COORDINATES OF ALL ATOMS IN TARGET MOLECULE

2200 CALCULATE BINDING POTENTIAL OF EACH ATOM IN TARGET MOLECULE

2300 IDENTIFY REGIONS WITH HIGHEST BINDING POTENTIAL

2400 IDENTIFY POTENTIAL BINDING SITES IN TERMS OF BINDING POTENTIAL

2500 CLASSIFY POTENTIAL BINDING SITES ACCORDING TO CHEMICAL NATURE (E.G., DEGREE OF HYDROPHOBICITY, CHARGE, ETC.), SURFACE AREA, LOCATION IN CAVITY, ETC.

2600 SELECT POTENTIAL BINDING SITE FOR LIGAND DESIGN ACCORDING TO USER CRITERIA, E.G.:
-TYPE OF LIGAND (E.G., PEPTIDE, ORGANIC MOLECULE)
-BINDING AFFINITY
-SIZE
-GEOMETRY

FIG.3

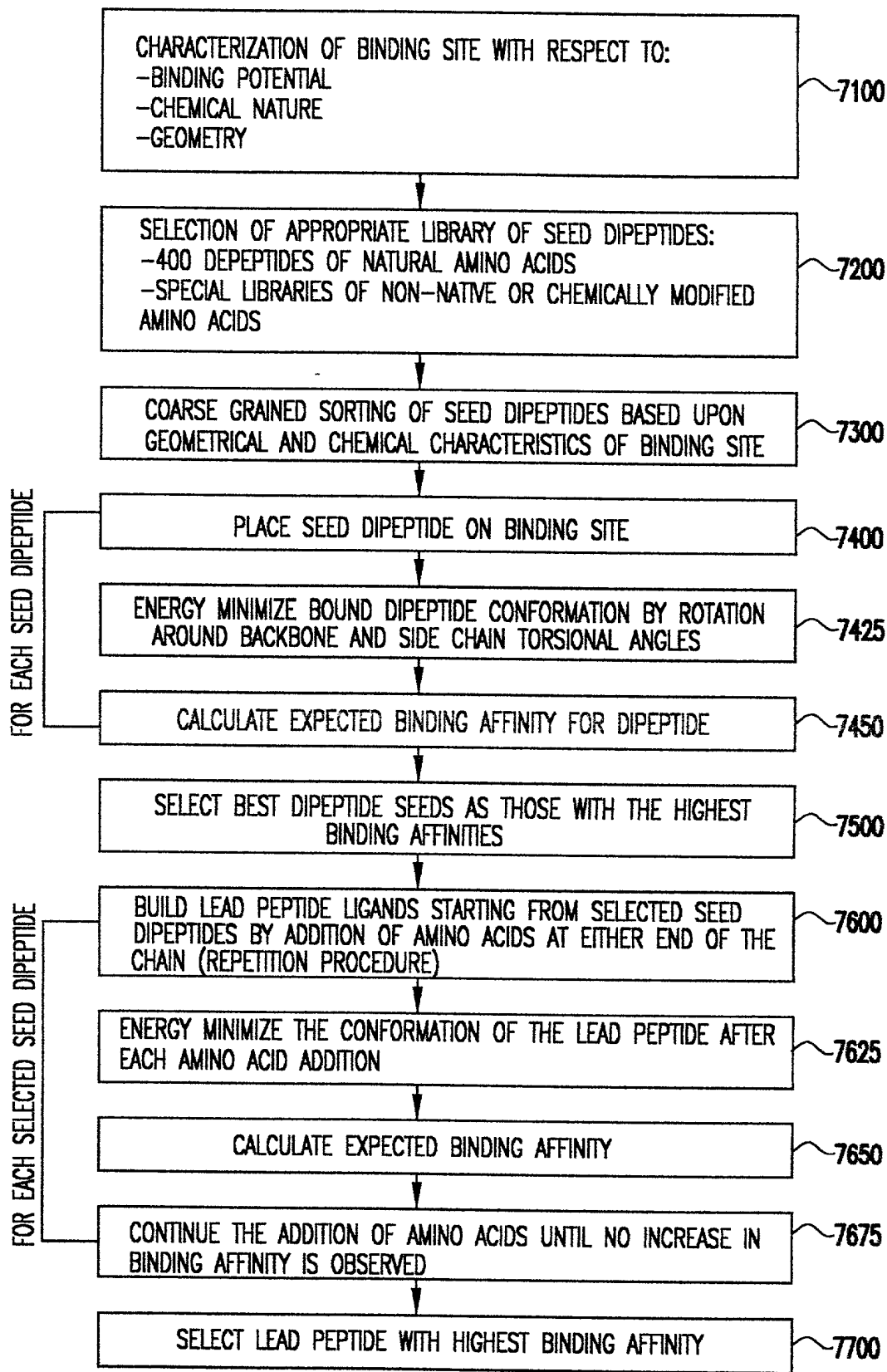


FIG.4

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graph TD; 6100[STRUCTURE-BASED THERMODYNAMIC ANALYSIS OF COMPLEX OF LEAD PEPTIDE WITH MACROMOLECULE] --> 6200[CALCULATION OF CONTRIBUTIONS OF INDIVIDUAL PEPTIDE RESIDUES OF BINDING ENERGETICS. CHANGES IN LEAD PEPTIDE WITH THE GOAL OF IMPROVED BINDING AFFINITY ARE USUALLY MADE AT THOSE POSITIONS THAT CONTRIBUTE THE LEAST TO GIBBS ENERGY OF BINDING]; 6200 --> 6300[MODIFICATION OF LEAD PEPTIDE IS ACCOMPLISHED BY A COMBINATION OF ONE OR MORE OF THE FOLLOWING OPERATIONS:  
1.) MUTATION  
2.) ADDITION OF AMINO ACIDS AT AMINO AND /OR CARBOXY TERMINUS OF LEAD PEPTIDE  
3.) DELETION OF AMINO ACIDS AT EITHER END OF LEAD PEPTIDE]; 6300 --> 6400["-AMINO ACIDS FOR MUTATION OR ADDITION ARE SELECTED FROM 20 NATURAL AMINO ACIDS, NON-NATURAL AMINO ACIDS OR CHEMICALLY MODIFIED AMINO ACIDS  
-AMINO ACIDS ARE SYSTEMATICALLY CHOSEN AND INSERTED AT DESIRED POSITION FOR ADDITION OR REPLACEMENT  
-ATOMIC COORDINATES OF MODIFIED COMPLEX ARE GENERATED"]; 6400 --> 6500[FOR EACH CHANGE OR ADDITION THE CONFORMATION WITH THE LOWEST ENERGY IS IDENTIFIED BY SYSTEMATIC ROTATION AROUND BACKBONE AND SIDE CHAIN TORSIONAL ANGLES]; 6500 --> 6600[EXPECTED BINDING CONSTANT IS CALCULATED]; 6600 --> 6700[3D MODEL(I.E., ATOMIC COORDINATES) OF MODIFIED PEPTIDE BOUND TO MACROMOLECULE IS BUILT.]; 6700 --> 6100;
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STRUCTURE-BASED THERMODYNAMIC ANALYSIS OF COMPLEX OF LEAD PEPTIDE WITH MACROMOLECULE 6100

CALCULATION OF CONTRIBUTIONS OF INDIVIDUAL PEPTIDE RESIDUES OF BINDING ENERGETICS. CHANGES IN LEAD PEPTIDE WITH THE GOAL OF IMPROVED BINDING AFFINITY ARE USUALLY MADE AT THOSE POSITIONS THAT CONTRIBUTE THE LEAST TO GIBBS ENERGY OF BINDING 6200

MODIFICATION OF LEAD PEPTIDE IS ACCOMPLISHED BY A COMBINATION OF ONE OR MORE OF THE FOLLOWING OPERATIONS:
1.) MUTATION
2.) ADDITION OF AMINO ACIDS AT AMINO AND /OR CARBOXY TERMINUS OF LEAD PEPTIDE
3.) DELETION OF AMINO ACIDS AT EITHER END OF LEAD PEPTIDE 6300

-AMINO ACIDS FOR MUTATION OR ADDITION ARE SELECTED FROM 20 NATURAL AMINO ACIDS, NON-NATURAL AMINO ACIDS OR CHEMICALLY MODIFIED AMINO ACIDS
-AMINO ACIDS ARE SYSTEMATICALLY CHOSEN AND INSERTED AT DESIRED POSITION FOR ADDITION OR REPLACEMENT
-ATOMIC COORDINATES OF MODIFIED COMPLEX ARE GENERATED 6400

FOR EACH CHANGE OR ADDITION THE CONFORMATION WITH THE LOWEST ENERGY IS IDENTIFIED BY SYSTEMATIC ROTATION AROUND BACKBONE AND SIDE CHAIN TORSIONAL ANGLES 6500

EXPECTED BINDING CONSTANT IS CALCULATED 6600

3D MODEL(I.E., ATOMIC COORDINATES) OF MODIFIED PEPTIDE BOUND TO MACROMOLECULE IS BUILT. 6700

FIG.5

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graph TD
    9100[CHARACTERIZATION OF BINDING SITE WITH RESPECT TO:  
-BINDING POTENTIAL  
-CHEMICAL NATURE  
-GEOMETRY] --> 9200[BUILD IDEAL LIGAND BY PLACING ATOMS THAT ARE ENERGETICALLY  
COMPLEMENTARY TO THOSE IN THE BINDING SITE AT OPTIMAL  
VAN DER WAALS DISTANCES]
    9200 --> 9300[SELECT LEAD COMPOUND USING GEOMETRICAL ARRANGEMENT OF  
ATOMS IN IDEAL LIGAND AS A TEMPLATE. SELECTION IS MADE FROM  
DATABASE OF COMPOUNDS OR BY BUILDING MOLECULE WITH  
APPROPRIATE ATOMIC ARRANGEMENT.]
    9300 --> 9400[PLACE LEAD COMPOUND ON BINDING SITE.]
    9400 --> 9425[ENERGY MINIMIZE BOUND CONFORMATION BY SYSTEMATIC BOND  
ROTATION.]
    9425 --> 9450[CALCULATE EXPECTED BINDING AFFINITY FOR LEAD COMPOUND]
    9450 --> 9500[SELECT BEST LEADS AS THOSE WITH THE HIGHEST BINDING  
AFFINITIES.]
    9500 --> 9600[OPTIMIZE LEAD COMPOUND LIGANDS BY ADDITION OR REPLACEMENT  
OF CHEMICAL GROUPS.]
    9600 --> 9625[ENERGY MINIMIZE THE CONFORMATION OF THE LEAD COMPOUND  
AFTER EACH MODIFICATION.]
    9625 --> 9650[CALCULATE EXPECTED BINDING AFFINITY.]
    9650 --> 9675[CONTINUE MODIFICATION UNTIL NO INCREASE IN BINDING AFFINITY IS  
OBSERVED.]
    9675 --> 9700[SELECT LEAD COMPOUND WITH HIGHEST BINDING AFFINITY.]

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FIG. 6

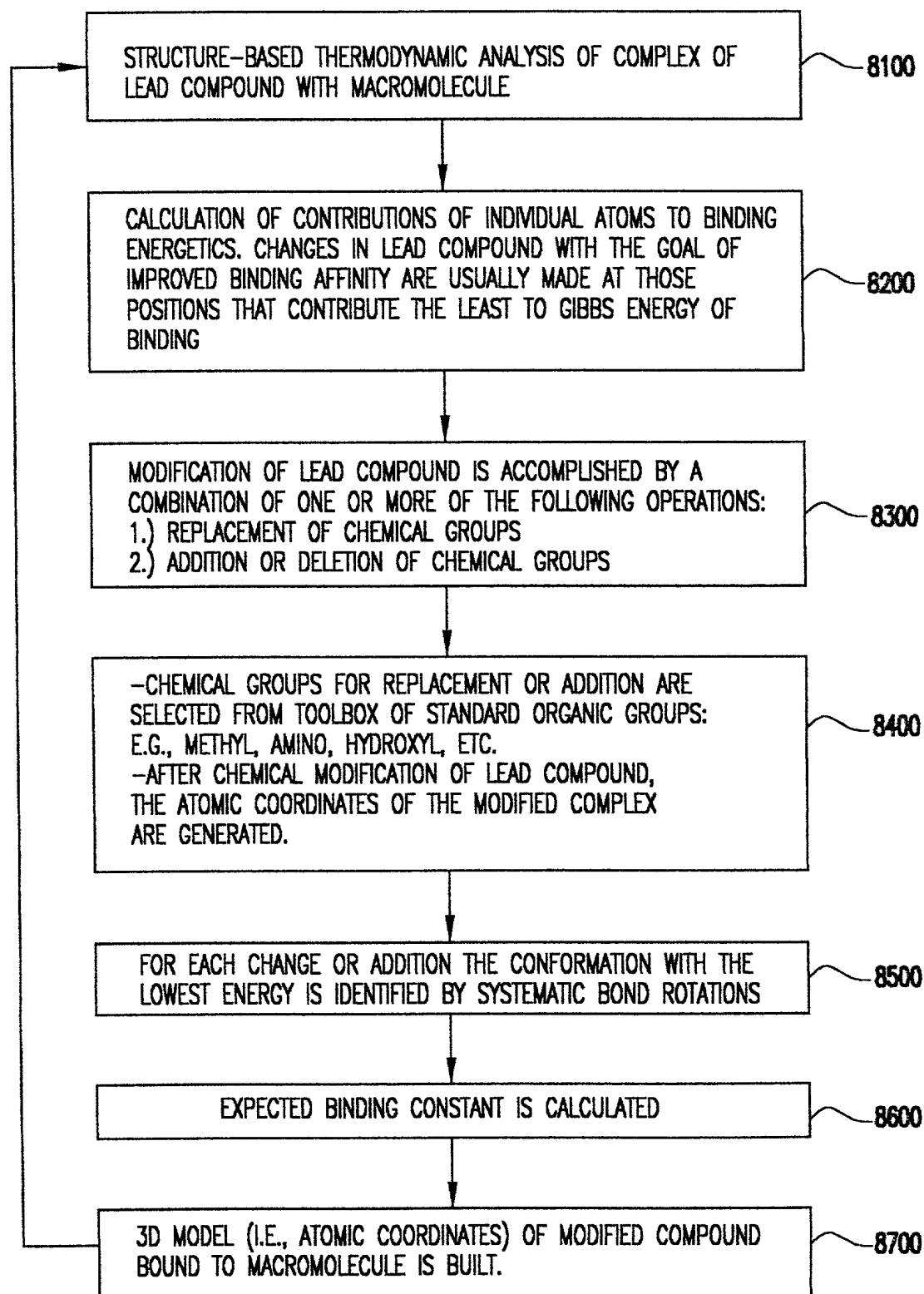


FIG.7

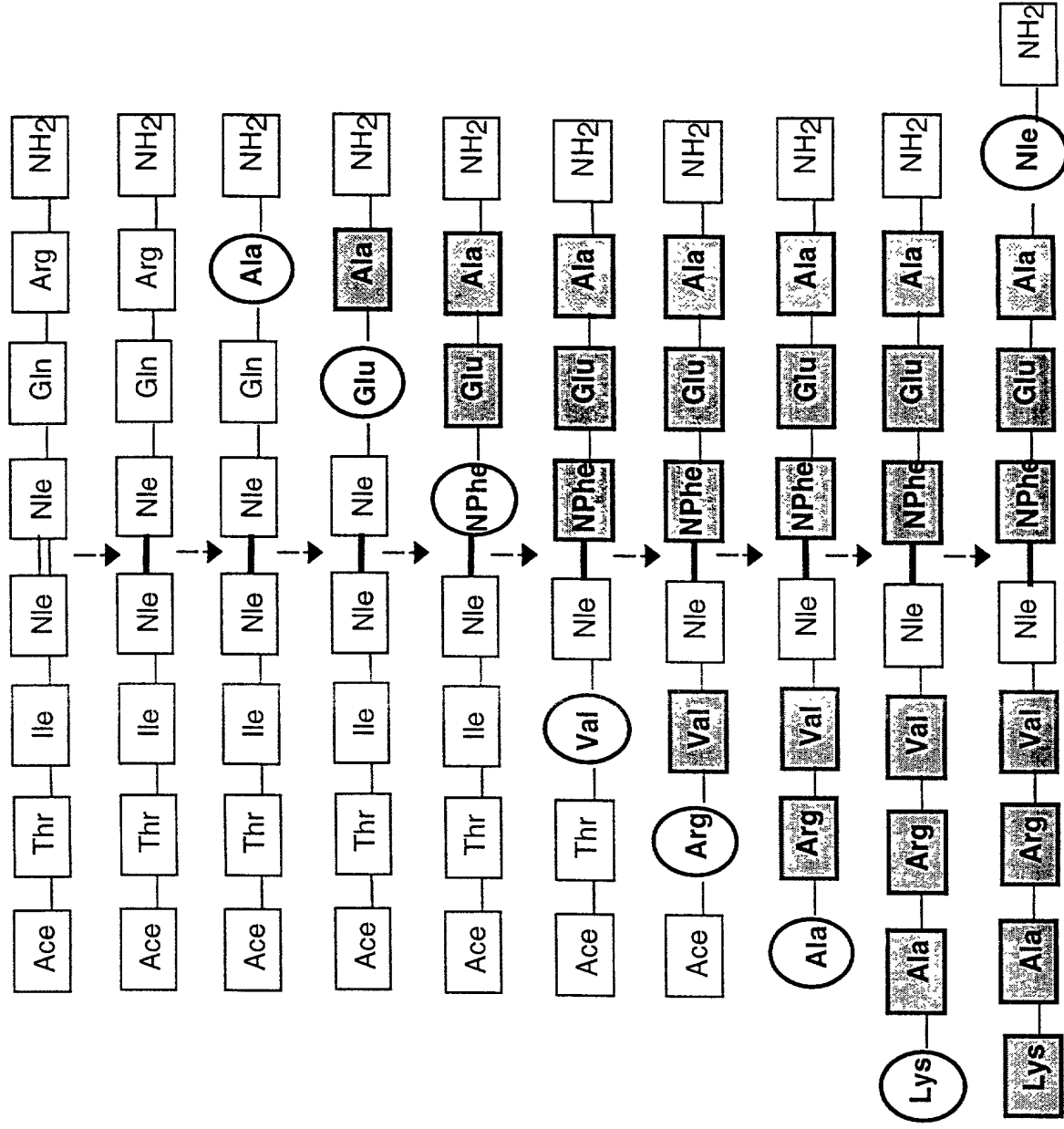
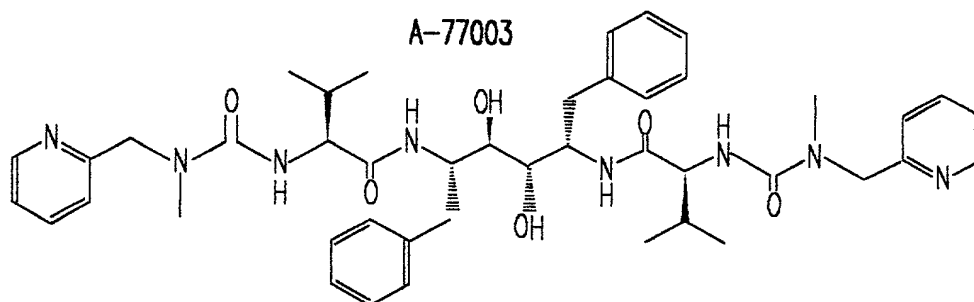
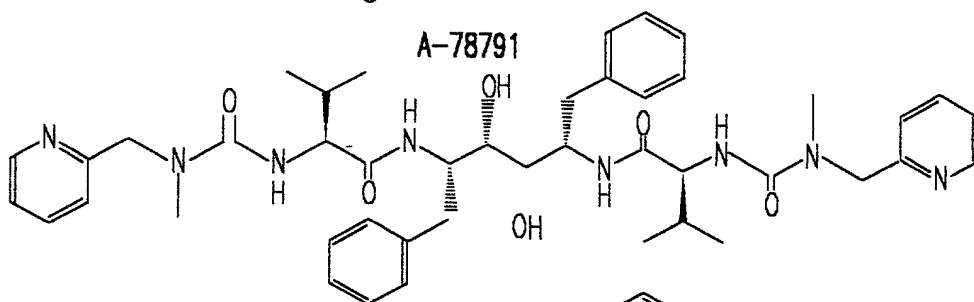


FIG.9

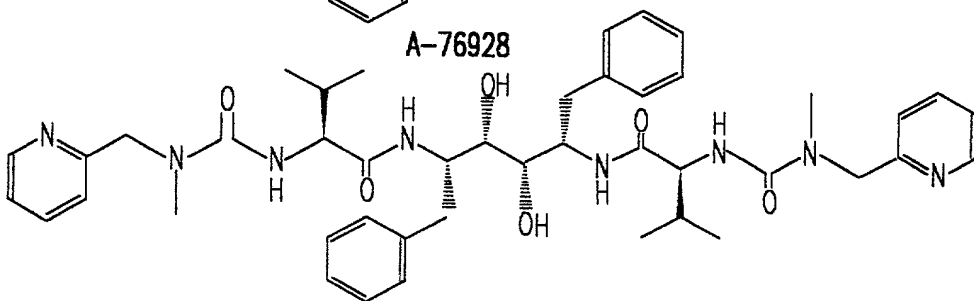
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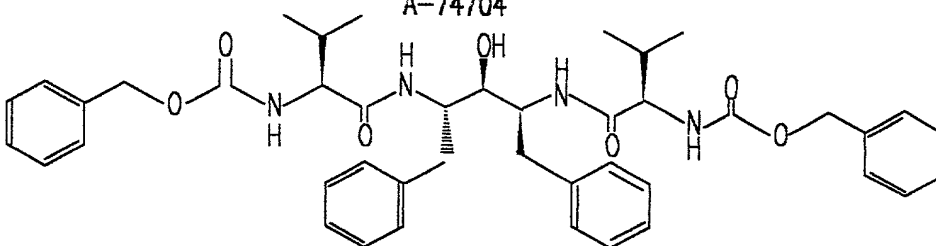
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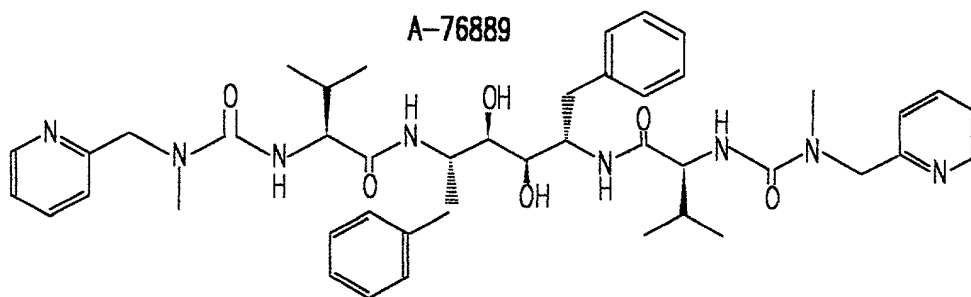
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A-74704



A-76889



VX478

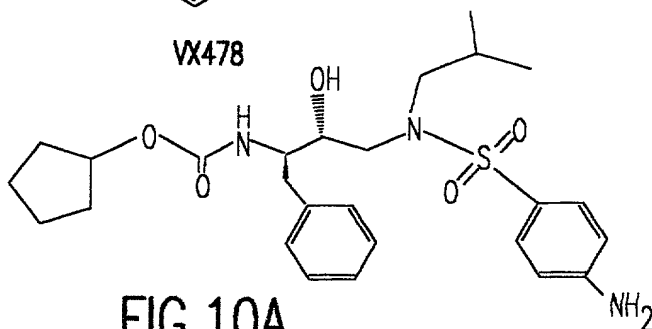
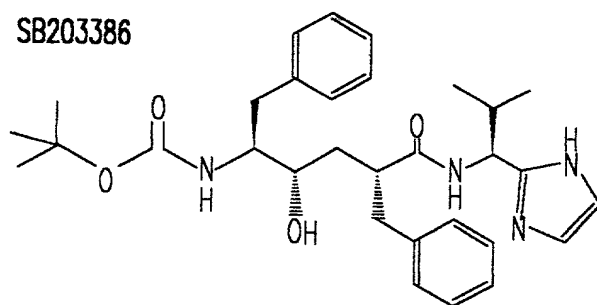
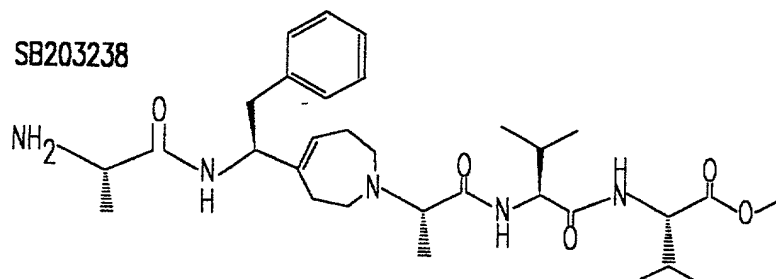


FIG. 10A

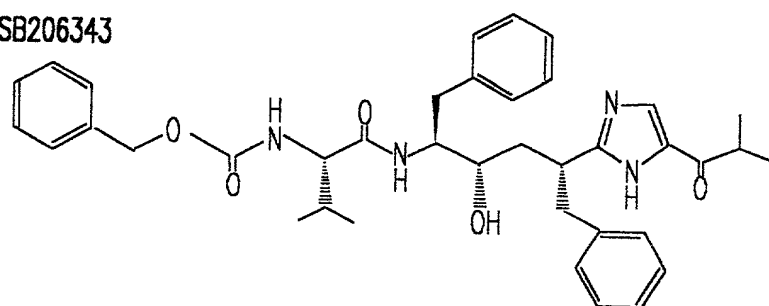
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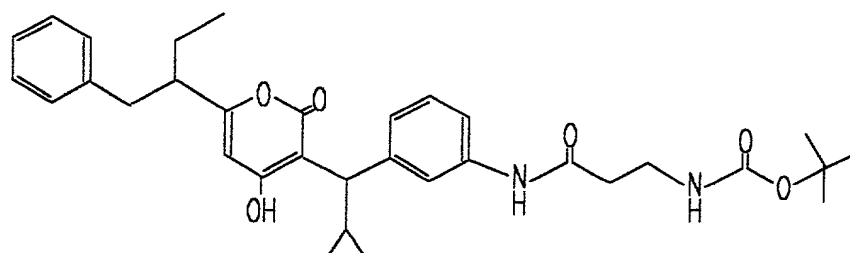
SB203238



SB206343



U100313



U-89360

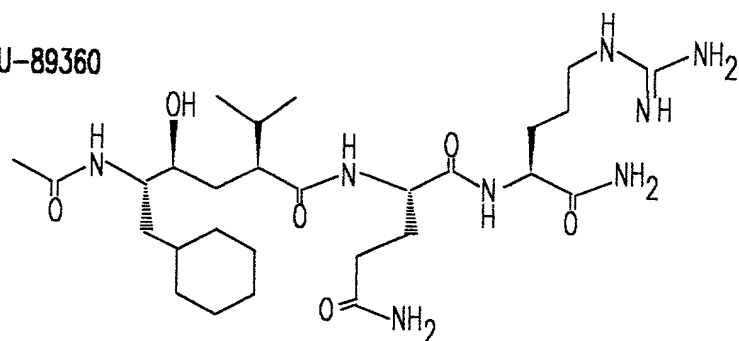
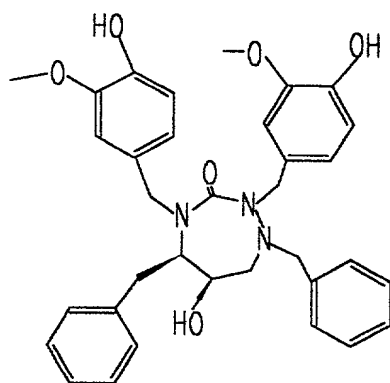


FIG.10B

A-98881



CGP 53820

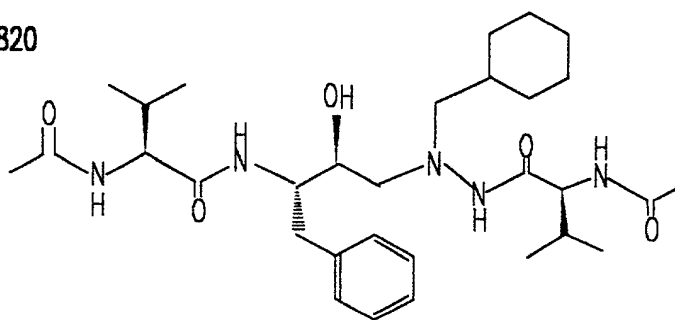


FIG.10C

00EYET*969HE260

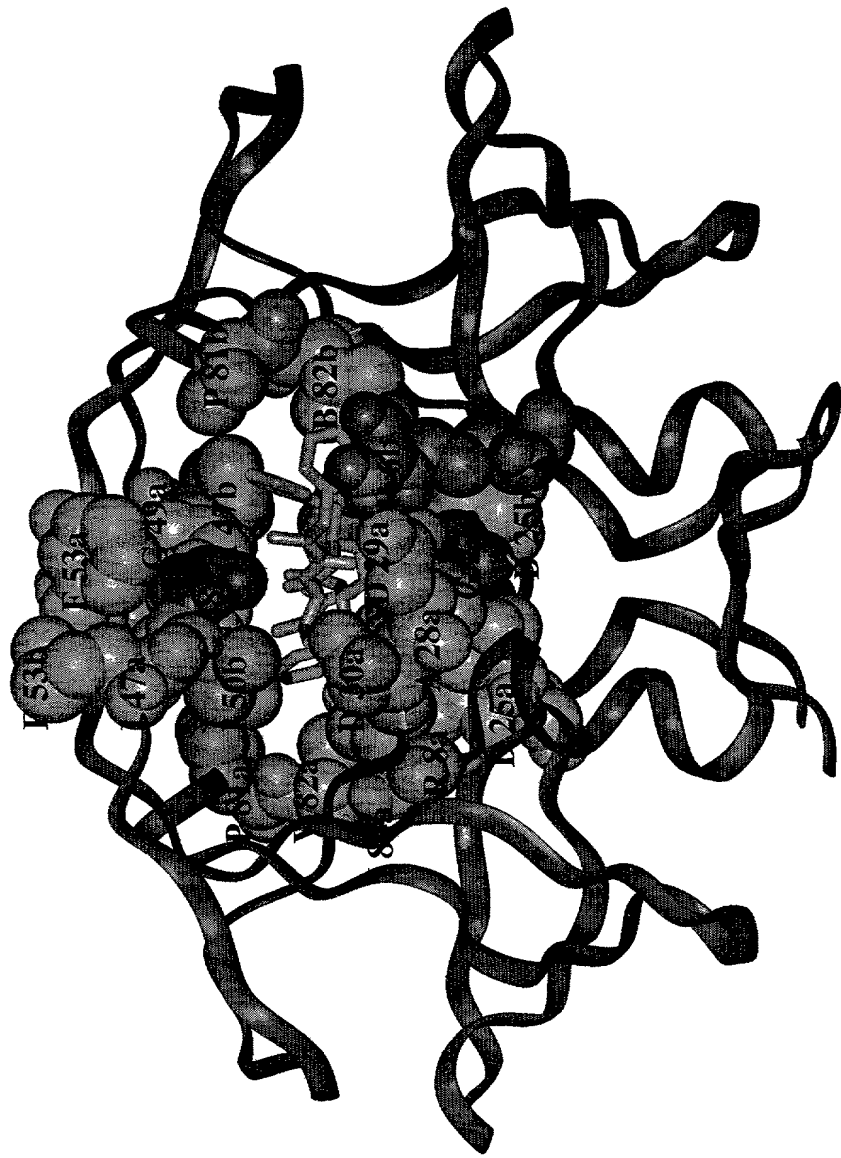


FIG.12A

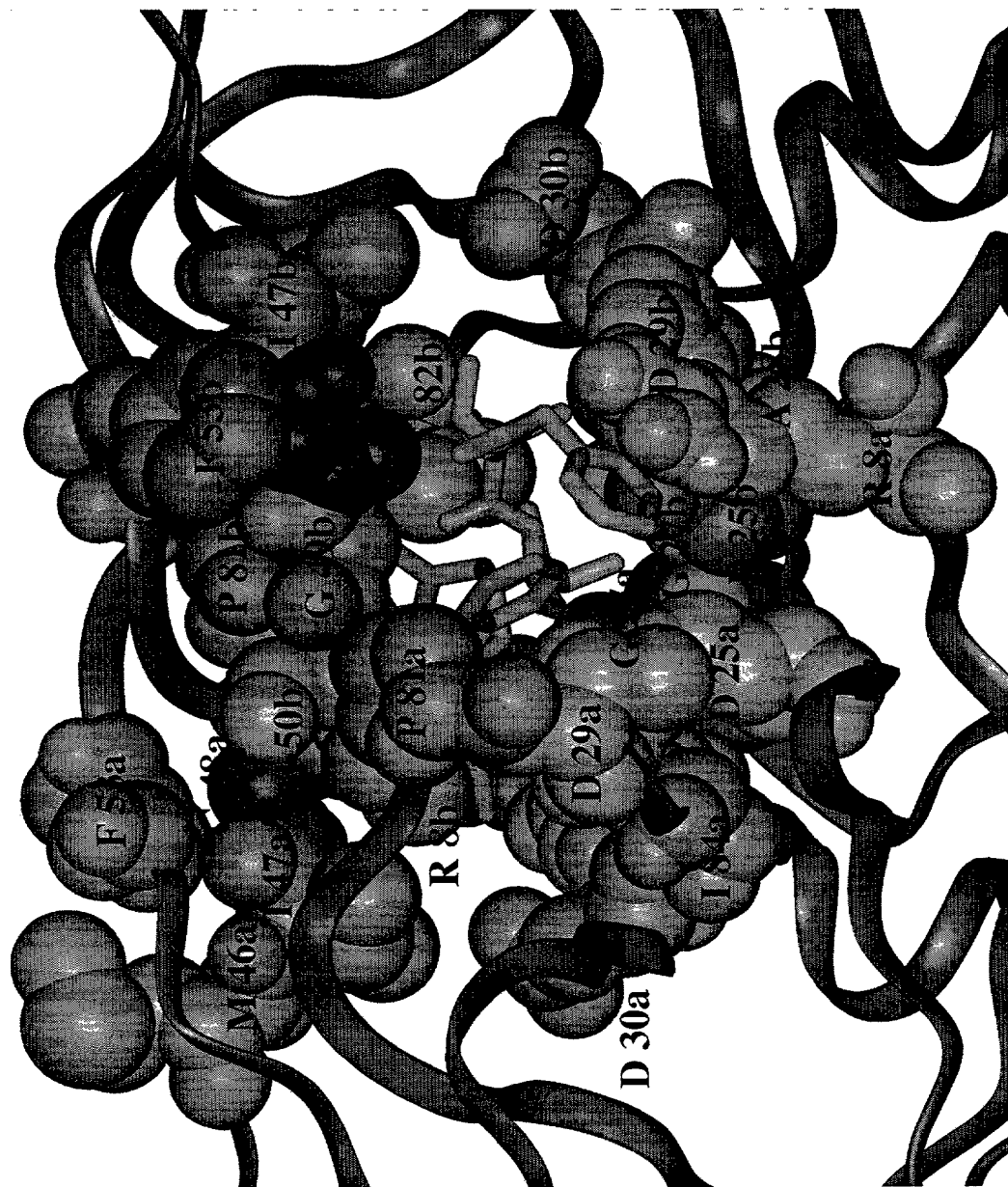


FIG.12B

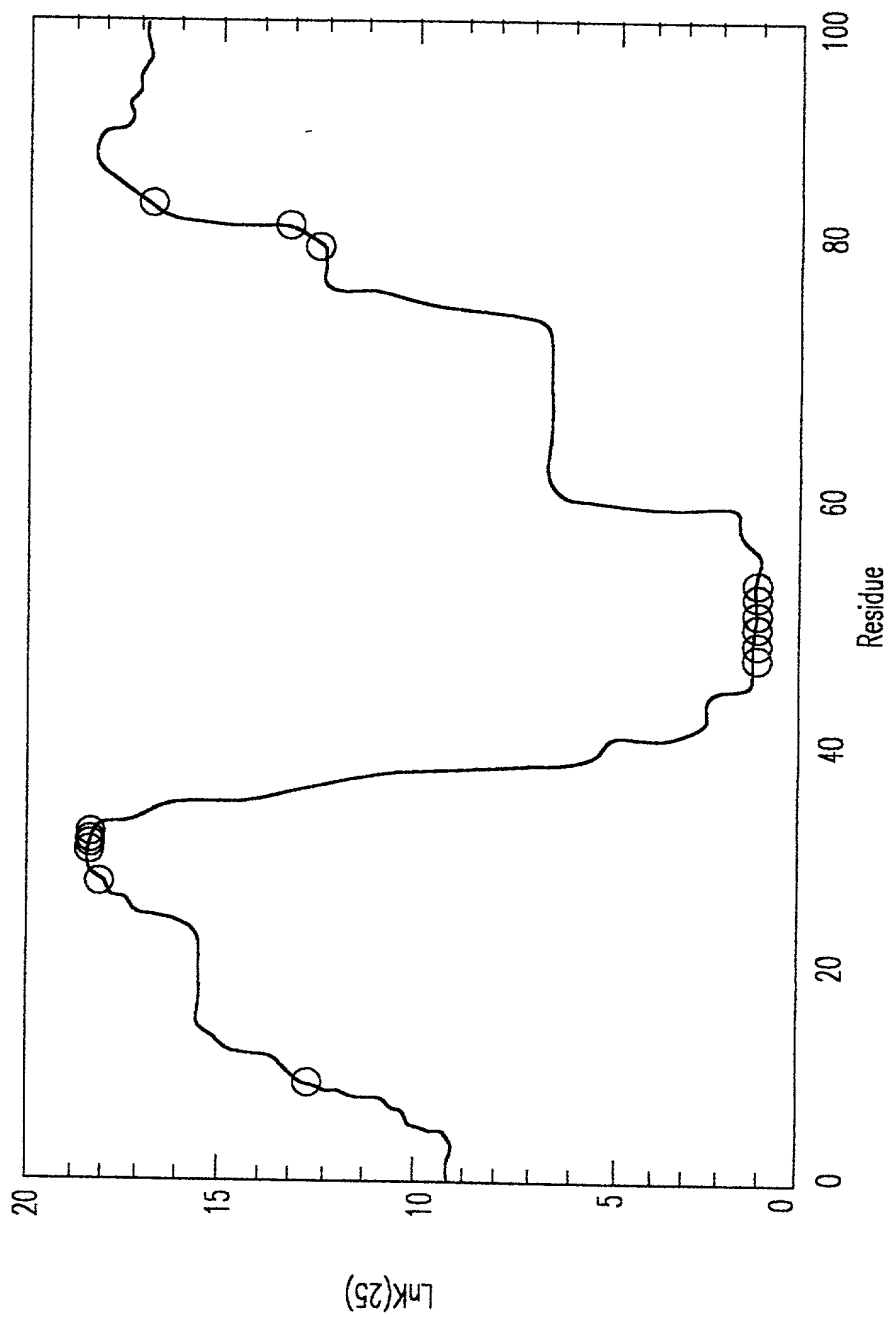


FIG.13

1 6
3 7

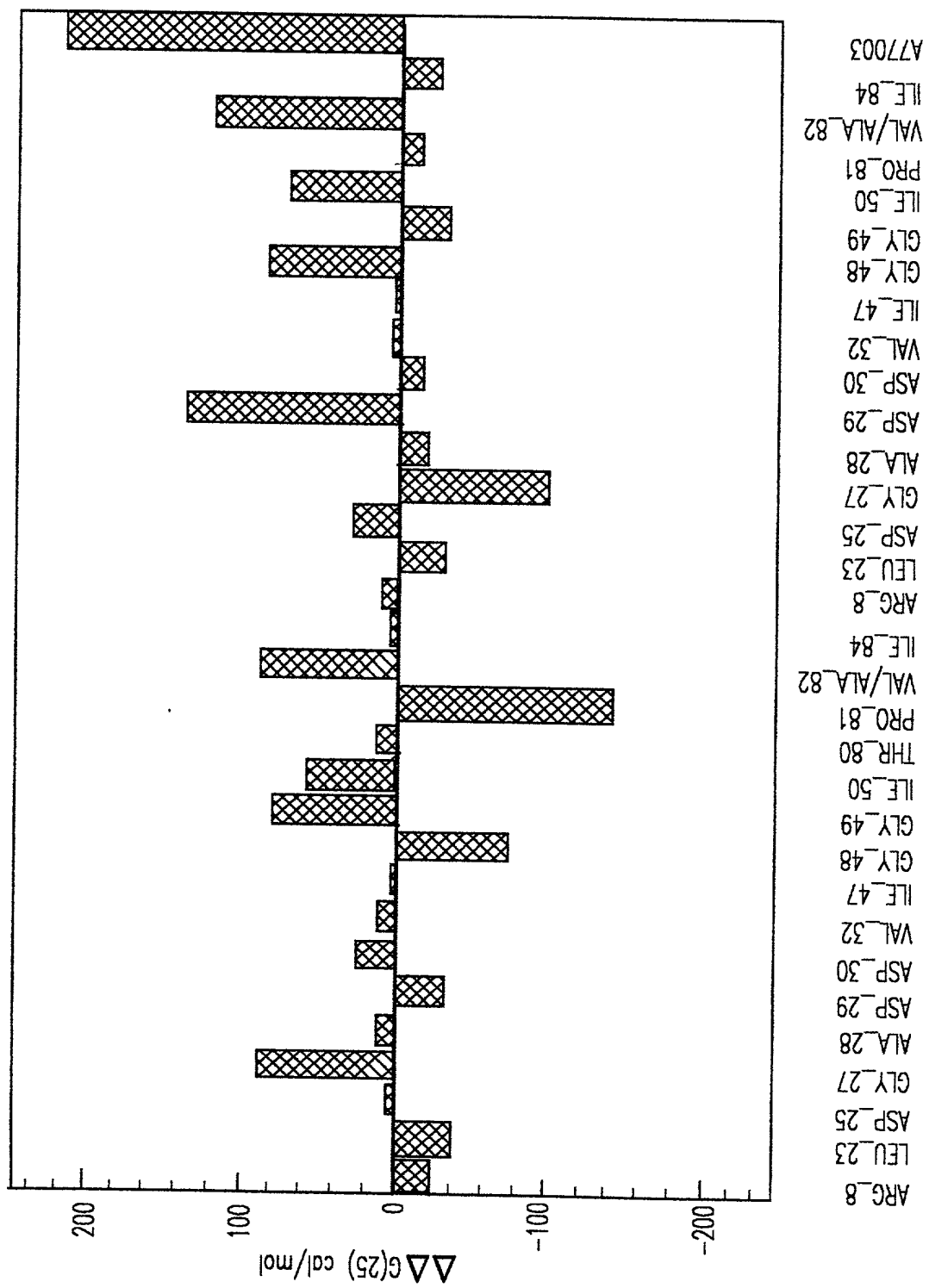


FIG. 14

00731596 124300

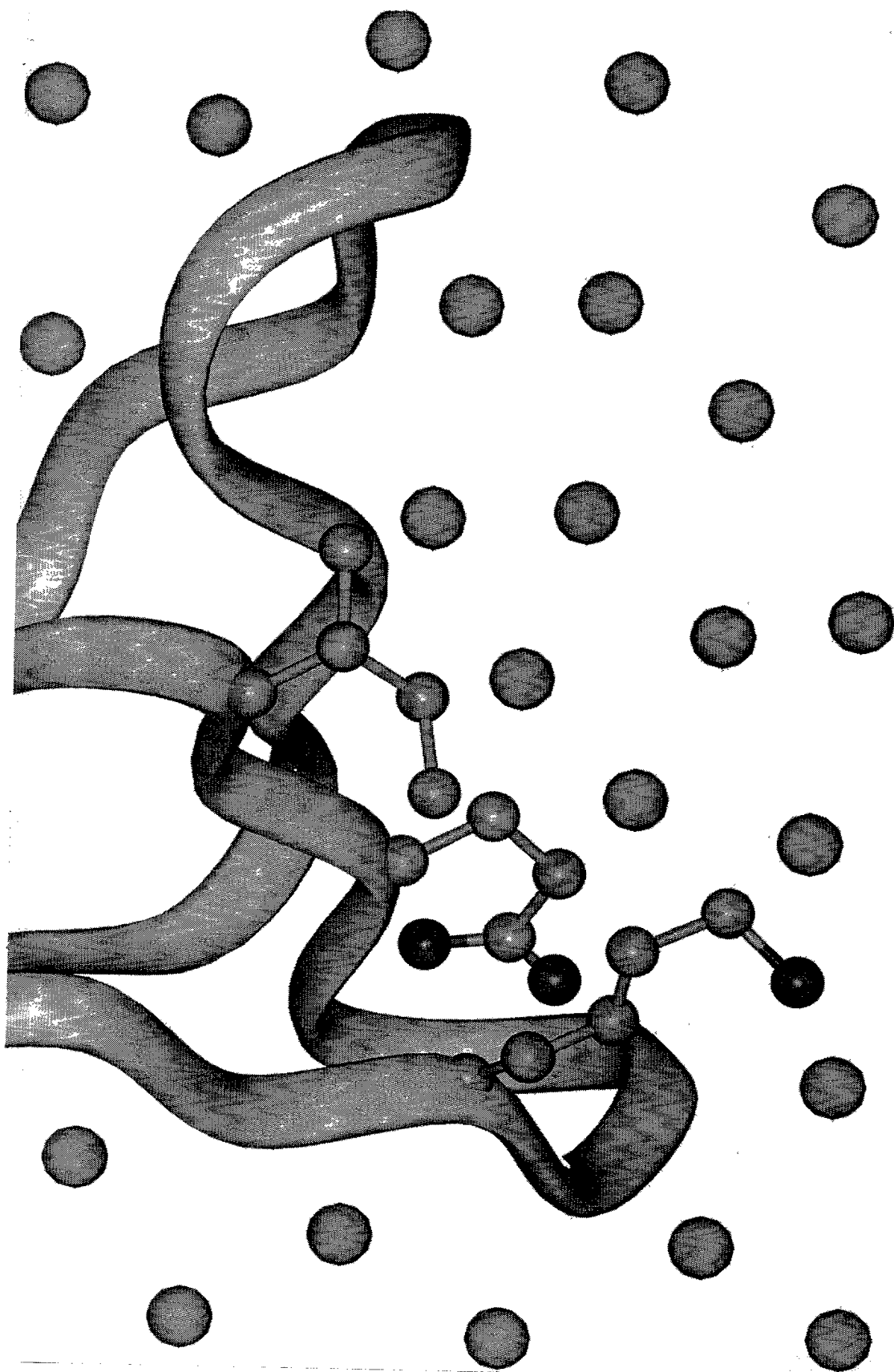


FIG.15A

FIG. 15B

X2

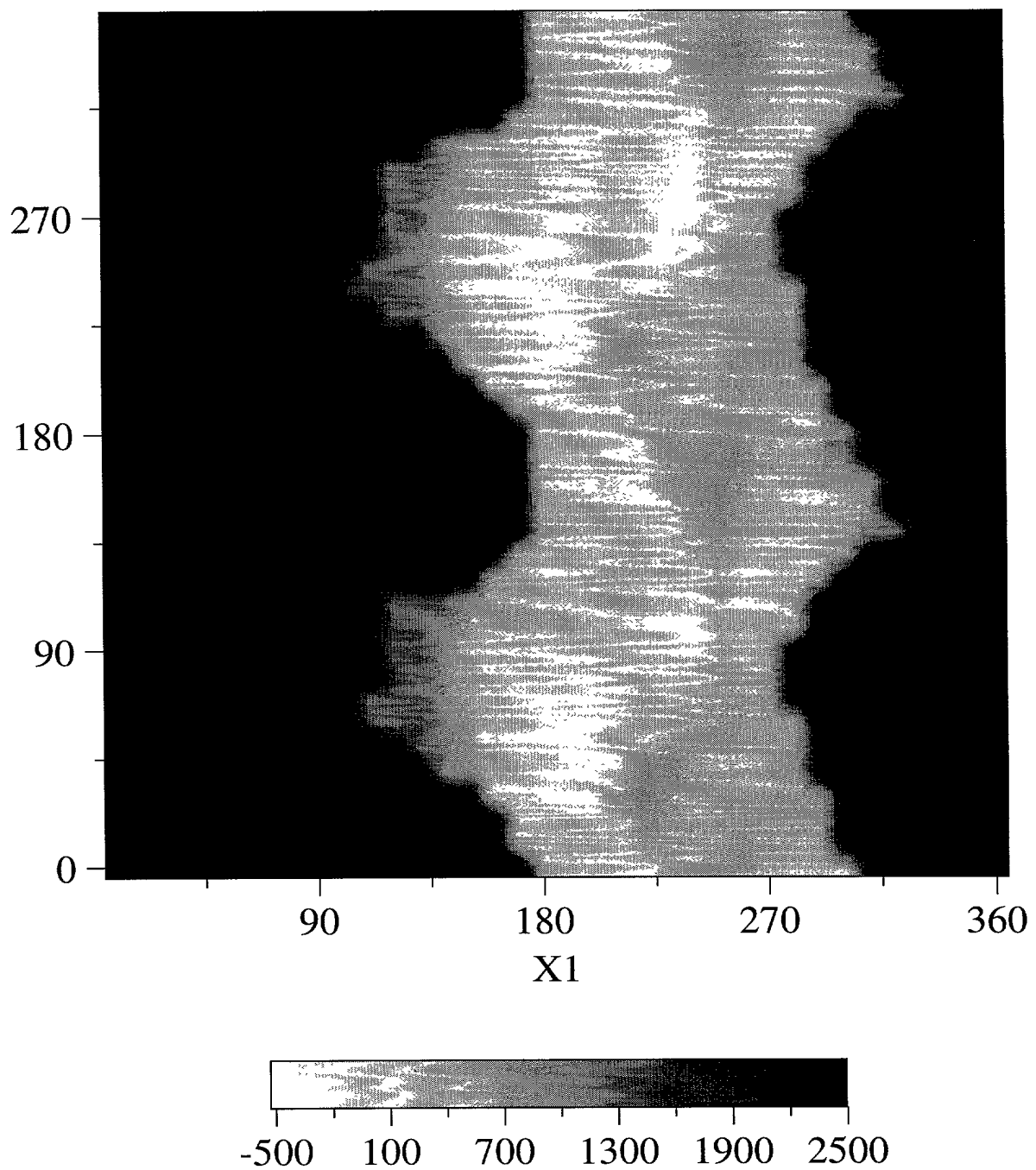
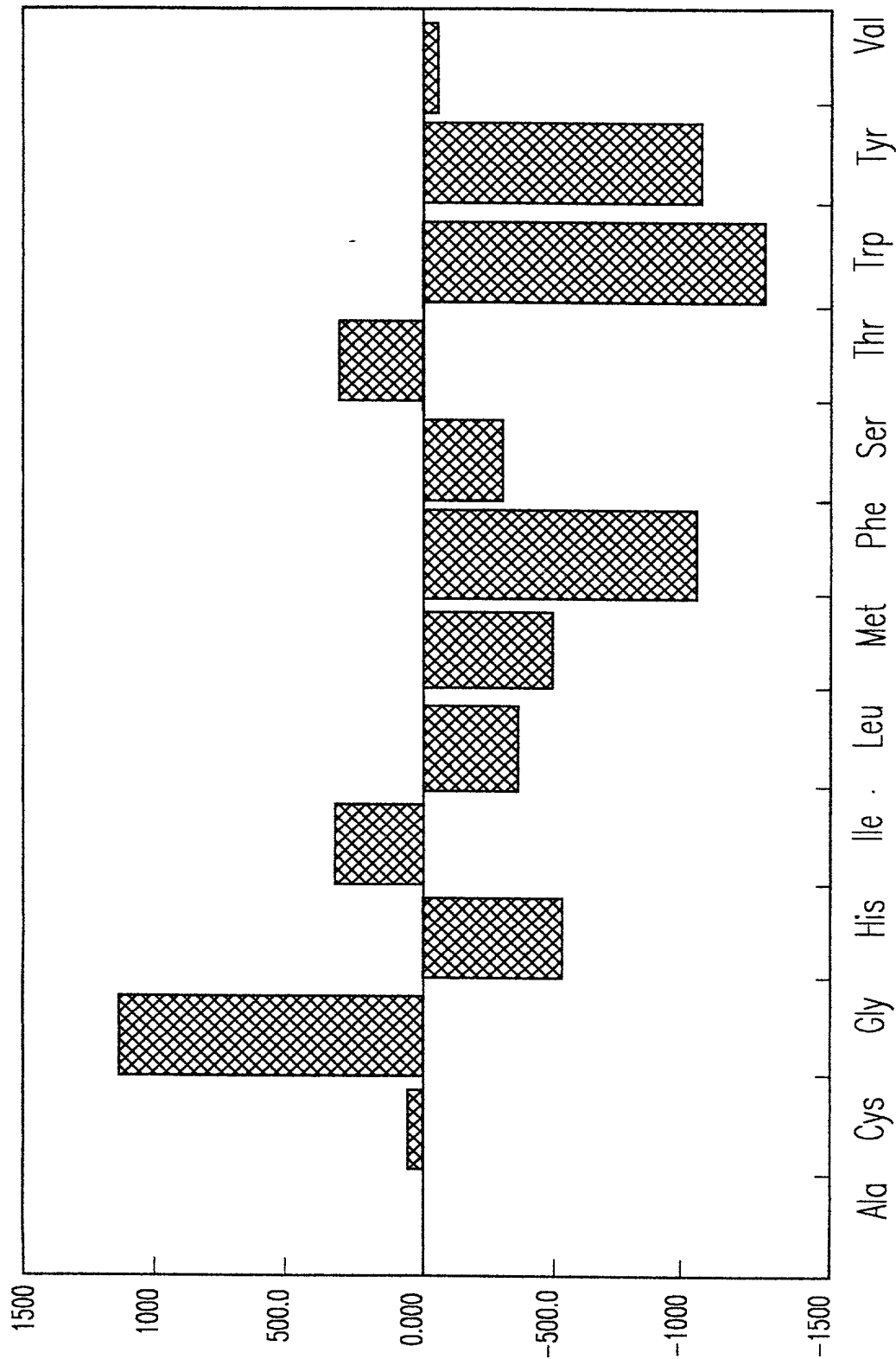


FIG. 16



Amino Acid

FIG.18

FIG. 19B

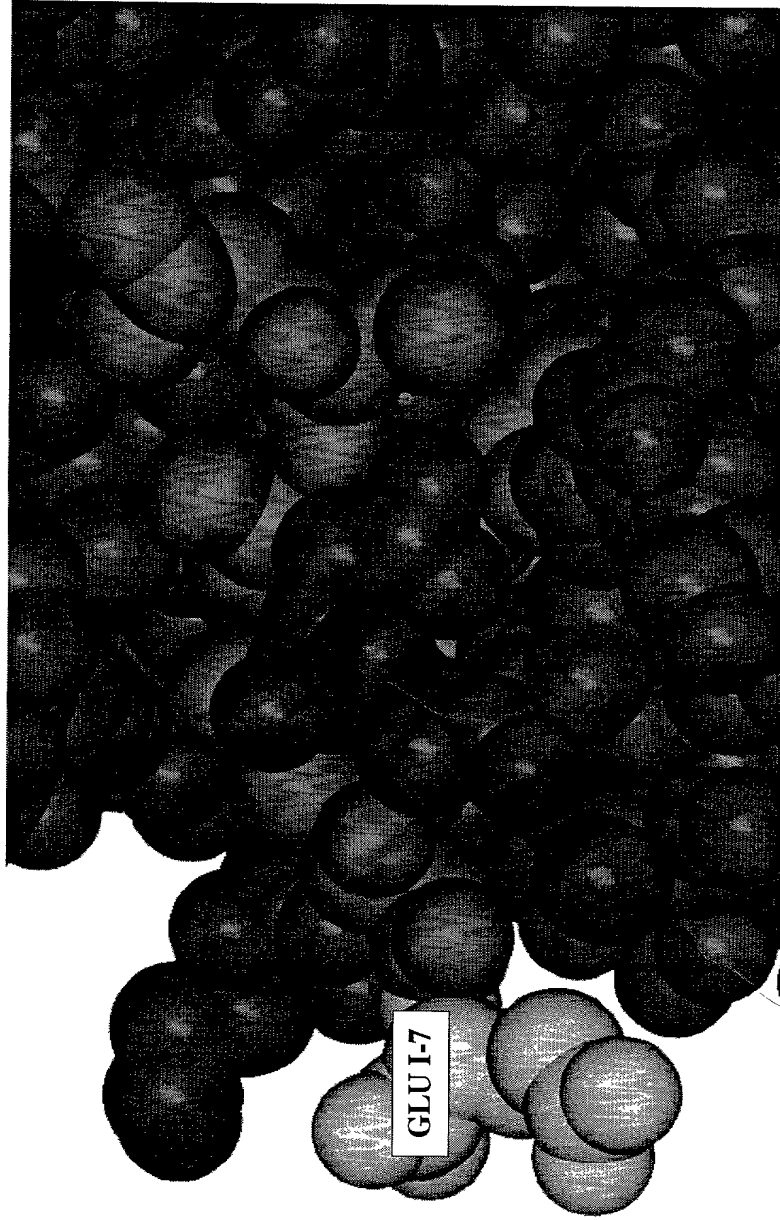


FIG.20